

# Is the molecule IR active?

① molecular shape & polar bonds



separation of charge



valence  $e^-$ 's =  $4 + 6 = 10$

↙ cloud of  $e^-$ 's



electronegativity - tendency for an atom to pull  $e^-$  density towards itself in a bond

# Homework

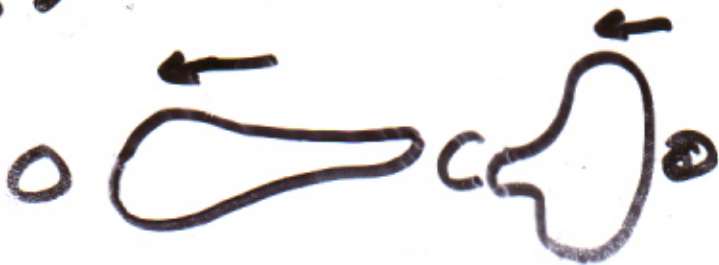
determine which vibrations  
are IR active for

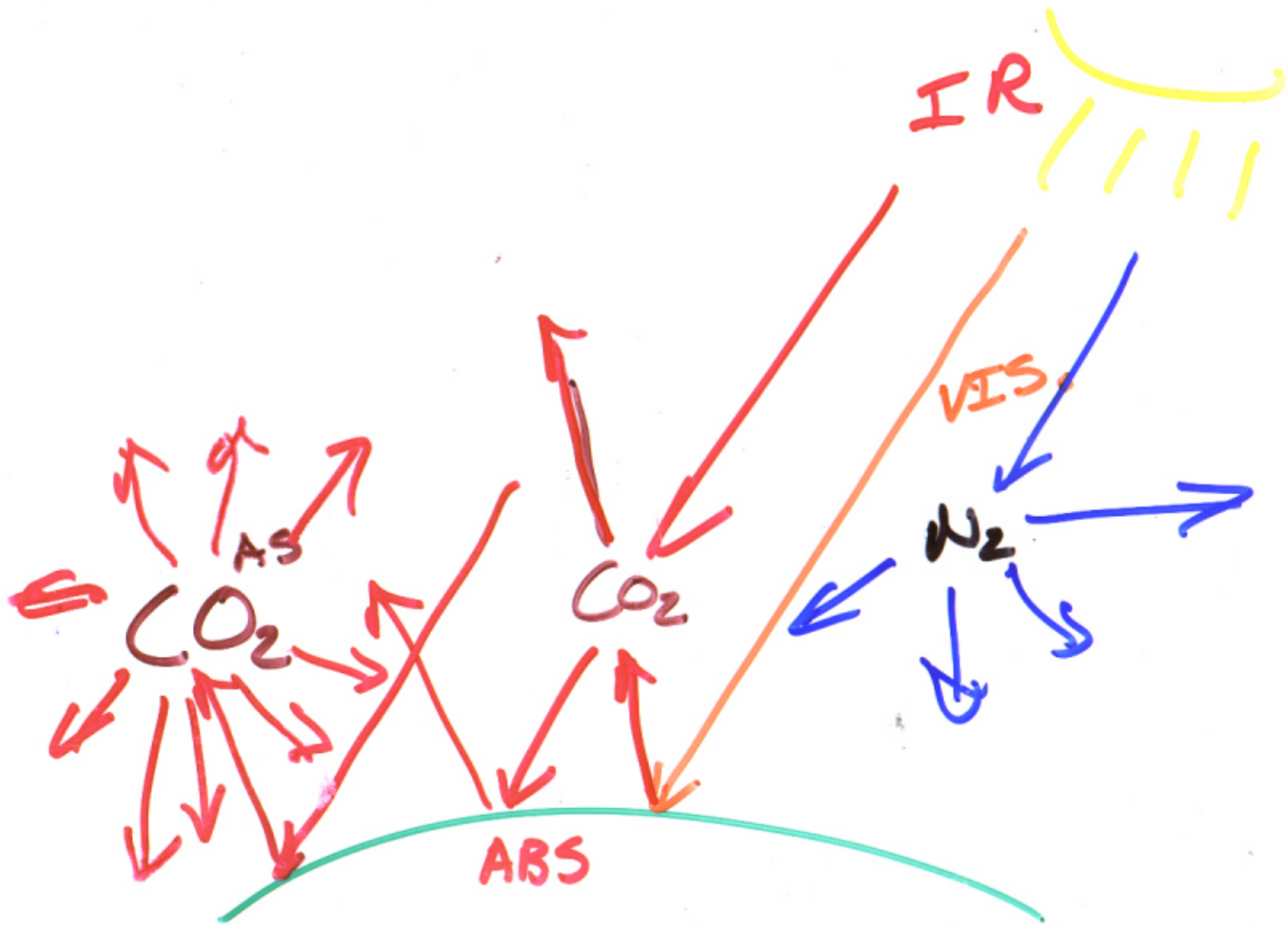
$\text{CO}_2$ ,  $\text{H}_2\text{O}$ ,  $\text{N}_2\text{O}$

do # 2 & 3 on p 27 of  
global warming module

summarize "what determines  
whether a gas is a  
greenhouse gas?"

Due week 5





$$0 = C = 0$$

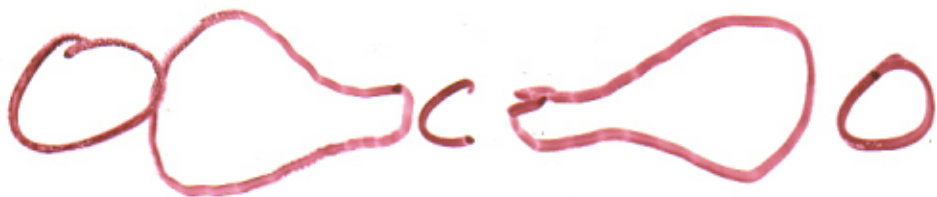
ASYM.



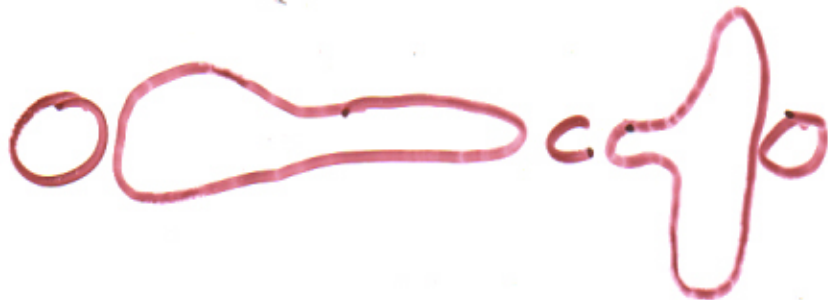
ADD VECTORS



NET CHANGE  
IN DIPOLE  
MOMENT

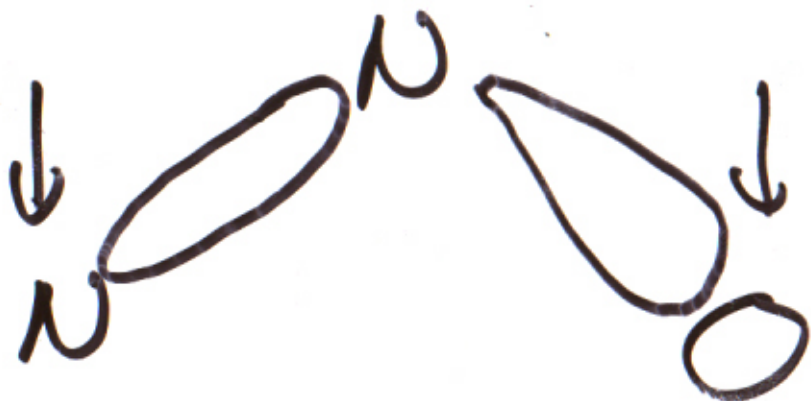
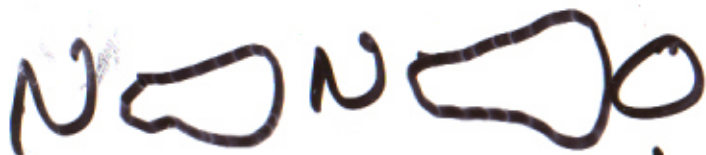
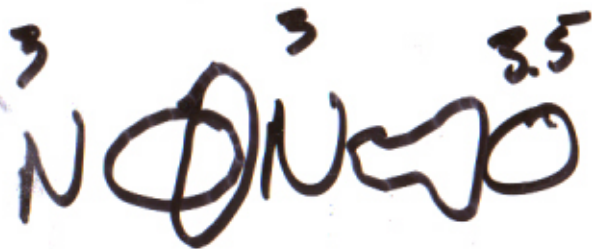


SYMMETRIC

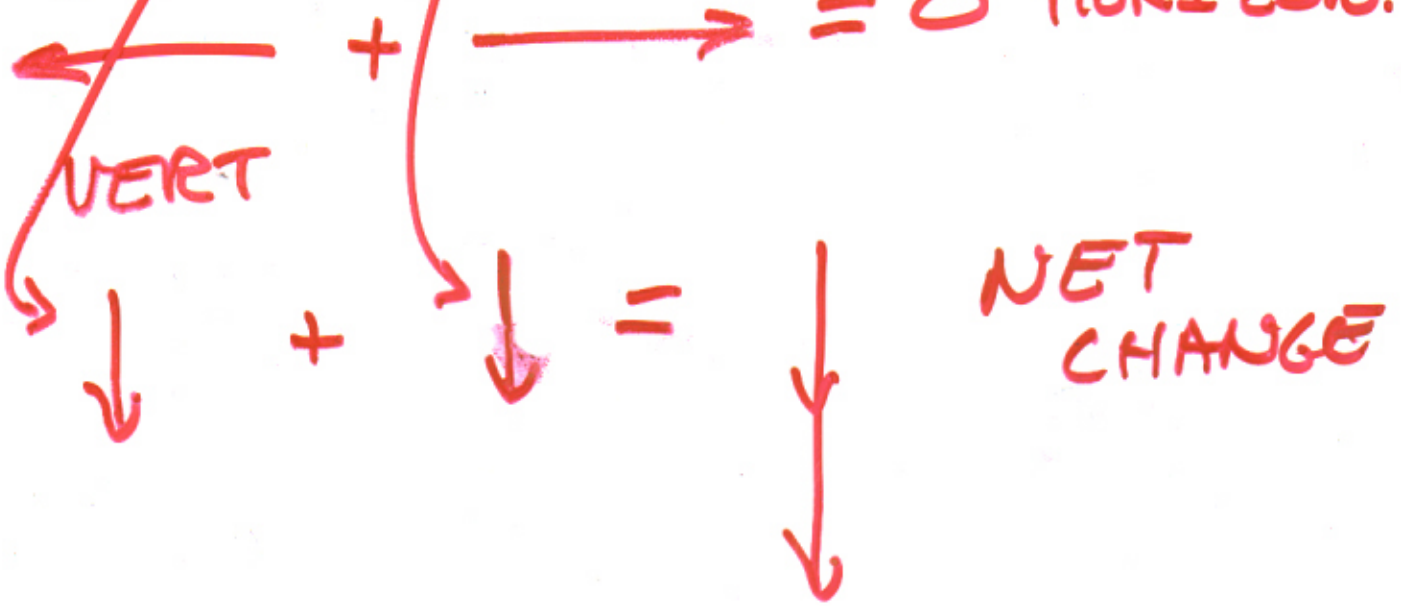
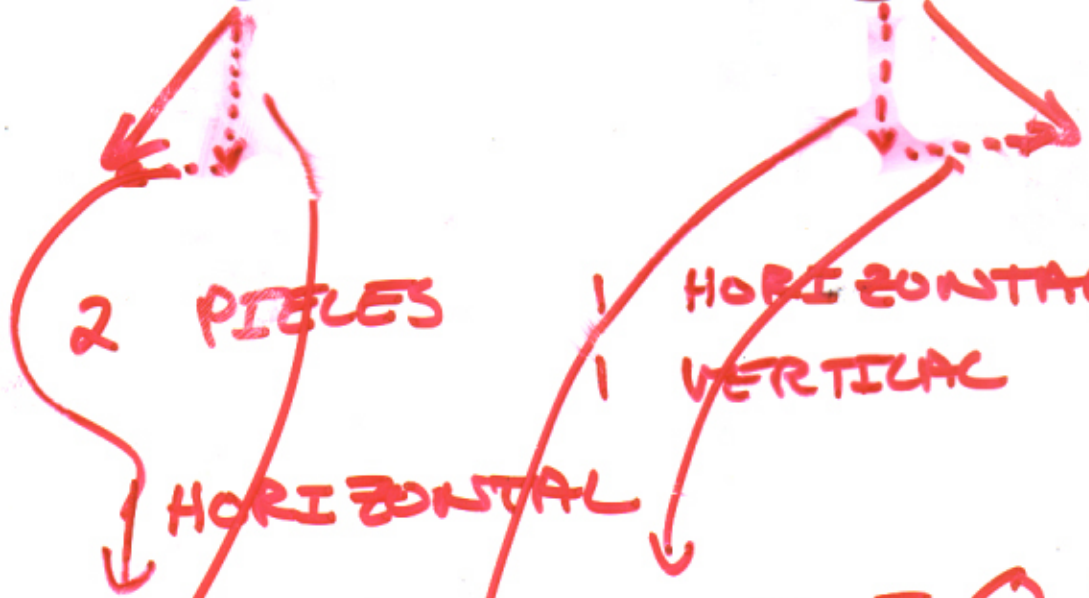


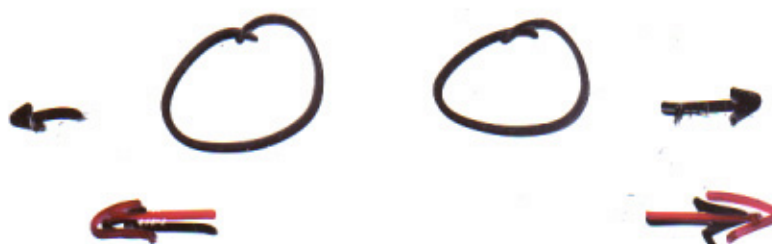
ASYM.

# DIPOLE MOMENT



O = C = O





$O_2 = 21\%$   
 $N_2 = 70\%$

VECTORS  
 • DIRECTION  
 • AMOUNT

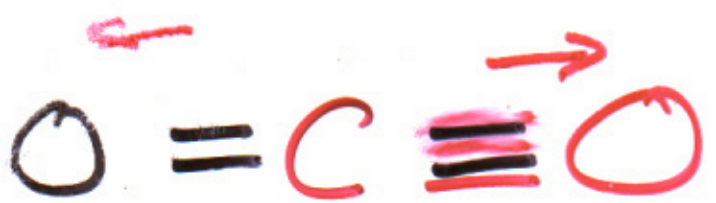
ADD VECTORS



TIP TO TAIL



NET CHANGE = 0



SYMMETRIK

ADD VECTORS



NO NET CHANGE

A molecule will absorb IR radiation during vibration if it has a change in the dipole moment during vibration

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nonpolar  
bond



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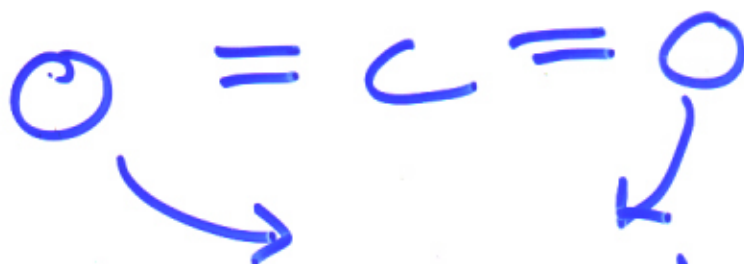
Sketch each molecular vibrations



symmetric stretch



asymmetric stretch



Bend



nonpolar  
bond



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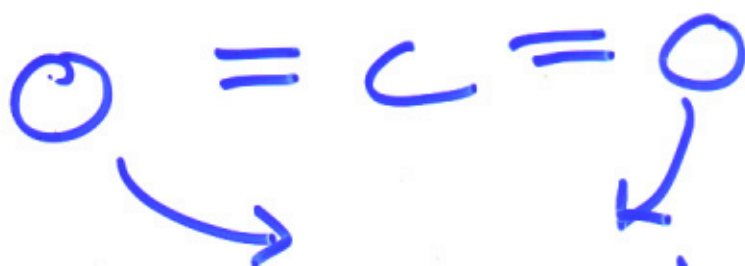
Sketch each molecular vibrations



symmetric stretch



asymmetric stretch



Bend