

ATOMS, MOLECULES & RESEARCH
COORDINATION CHEMISTRY - SPRING - WEEK 0

Generate $\Pi_{t_{2g}^2}$, $\Pi_{t_{2g}e_g}$ and $\Pi_{e_g^2}$ and reduce them to irreducible components in the O_h point group.

O_h	E	$8C_3$	$6C_2$	$6C_4$	$3C_2$	i	$6S_4$	$8S_6$	$3\sigma_h$	$6\sigma_d$
T_{2g}	3	0	1	-1	-1	3	-1	0	-1	1
E_g	2	-1	0	0	2	2	0	-1	2	0
$\Pi_{t_{2g}^2} =$	9	0	1	1	1	9	1	0	1	1
$\Pi_{t_{2g}e_g} =$	6	0	0	0	-2	6	0	0	-2	0
$\Pi_{e_g^2} =$	4	1	0	0	4	4	0	1	4	0

of times A_{1g} is in $\Pi_{t_{2g}^2} = \frac{1}{48} [4 + 8 + 12 + 4 + 8 + 12] = 1$

of times T_{2g} is in $\Pi_{t_{2g}^2} = \frac{1}{48} [27 + 6 - 6 - 3 + 27 - 6 + 6] = 1$

of times T_{1g} is in $\Pi_{t_{2g}^2} = \frac{1}{48} [27 - 6 + 6 - 3 + 27 + 6 - 3 - 6] = 1$

of times E_g is in $\Pi_{t_{2g}^2} = \frac{1}{48} [18 + 6 + 18 + 6] = 1$

$\Pi_{t_{2g}^2} = A_{1g} + T_{2g} + T_{1g} + E_g$

$$\left. \begin{array}{l} \# \text{ of times } T_{1g} \text{ in } \Pi \\ \Pi_{t_{2g}-e_g} \end{array} \right\} = \frac{1}{48} [18 + 6 + 18 + 6] = 1$$

$$\left. \begin{array}{l} \# \text{ of times } T_{2g} \text{ in } \Pi \\ \Pi_{t_{2g}-e_g} \end{array} \right\} = \frac{1}{48} [18 + 6 + 18 + 6] = 1$$

$$\underline{\underline{\Pi_{t_{2g}-e_g} = T_{1g} + T_{2g}}}}$$

$$\left. \begin{array}{l} \# \text{ of times } A_{1g} \text{ in } \Pi \\ \Pi_{e_g^2} \end{array} \right\} = \frac{1}{48} [4 + 8 + 12 + 4 + 8 + 12] = 1$$

$$\# \text{ of times } A_{2g} \text{ in } \Pi_{e_g^2} = \frac{1}{48} [4 + 8 + 12 + 4 + 8 + 12] = 1$$

$$\# \text{ of times } E_g \text{ in } \Pi_{e_g^2} = \frac{1}{48} [8 + 8 + 24 + 8 + 8 + 24] = 1$$

$$\underline{\underline{\Pi_{e_g^2} = A_{1g} + A_{2g} + E_g}}}$$

Chapter 11

- (a) $\frac{1}{T} \frac{1}{T} \frac{1}{T} e_g$ (b) $\frac{1}{A} \frac{1}{A} \frac{1}{A} e_g$ (c) $\frac{1}{E} \frac{1}{E} \frac{1}{E} e_g$
- $\frac{1}{T} \frac{1}{T} \frac{1}{T} t_{2g}$ $\frac{1}{A} \frac{1}{A} \frac{1}{A} t_{2g}$ $\frac{1}{E} \frac{1}{E} \frac{1}{E} t_{2g}$
- (d) $\frac{1}{T} \frac{1}{T} \frac{1}{T} e_g$ (e) $\frac{1}{E} \frac{1}{E} \frac{1}{E} e_g$
- $\frac{1}{T} \frac{1}{T} \frac{1}{T} t_{2g}$ $\frac{1}{E} \frac{1}{E} \frac{1}{E} t_{2g}$

(1) Sc^{3+} d^0 case } No Jahn-Teller distortion
 closed shell

$[Ti(NH_3)_6]^{3+}$ Ti^{3+} d^1 case } gr-state is degenerate.
 — — e_g }
 1 — — t_{2g} }
 gr-state = $\underline{\underline{T_{2g}}}$ }
 °. Jahn-Teller distortions possible but weak

$[V(NH_3)_6]^{3+}$ d^2 — — E_g } Ground state is degenerate.
 1 1 — — t_{2g} } $\underline{\underline{T_{2g}}}$ }
 Jahn-Teller distortions are possible, but weak

$[Cr(NH_3)_6]^{3+}$ d^3 — — } Ground state is not degenerate. No Jahn-Teller distortions possible.
 1 1 1 } $\underline{\underline{A}}$ }

$[Mn(NH_3)_6]^{3+}$ d^4 1 — — e_g } ground state is degenerate - strong Jahn-Teller distortions.
 (assume high spin) 1 1 1 — — t_{2g} } $\underline{\underline{E}}$ }

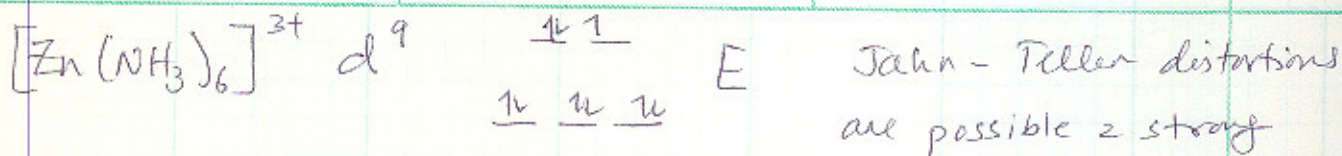
$[Fe(NH_3)_6]^{3+}$ d^5 1 1 } ground state is not degenerate - No Jahn-Teller distortions.
 1 1 1 — — } $\underline{\underline{A}}$ }

(~~high~~ low spin d^5 will have have Jahn-Teller distortions)

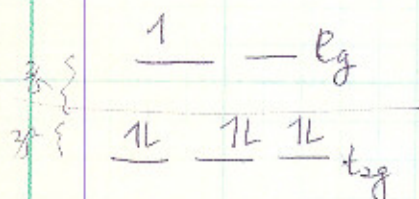
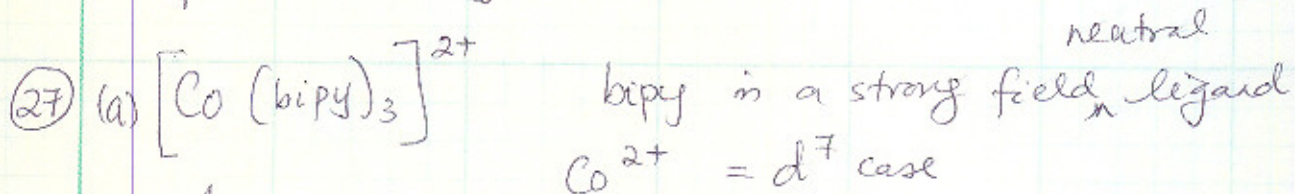
$[Co(NH_3)_6]^{3+}$ d^6 1 1 } ground state is degenerate. Jahn-Teller distortions are possible.
 1 1 1 — — } $\underline{\underline{T}}$ }

$[Ni(NH_3)_6]^{3+}$ d^7 1 1 }
 1 1 1 — — } $\underline{\underline{T}}$ }

$[Cu(NH_3)_6]^{3+}$ d^8 1 1 } No Jahn-Teller distortions.
 1 1 1 — — } $\underline{\underline{A}}$ }

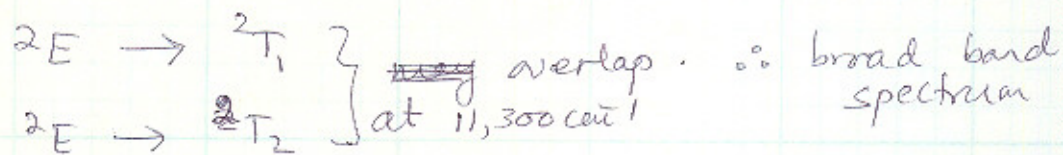


[Jahn-Teller distortions are stronger for E terms than for T terms].



Ground state term = $\underline{\underline{2E_g}}$

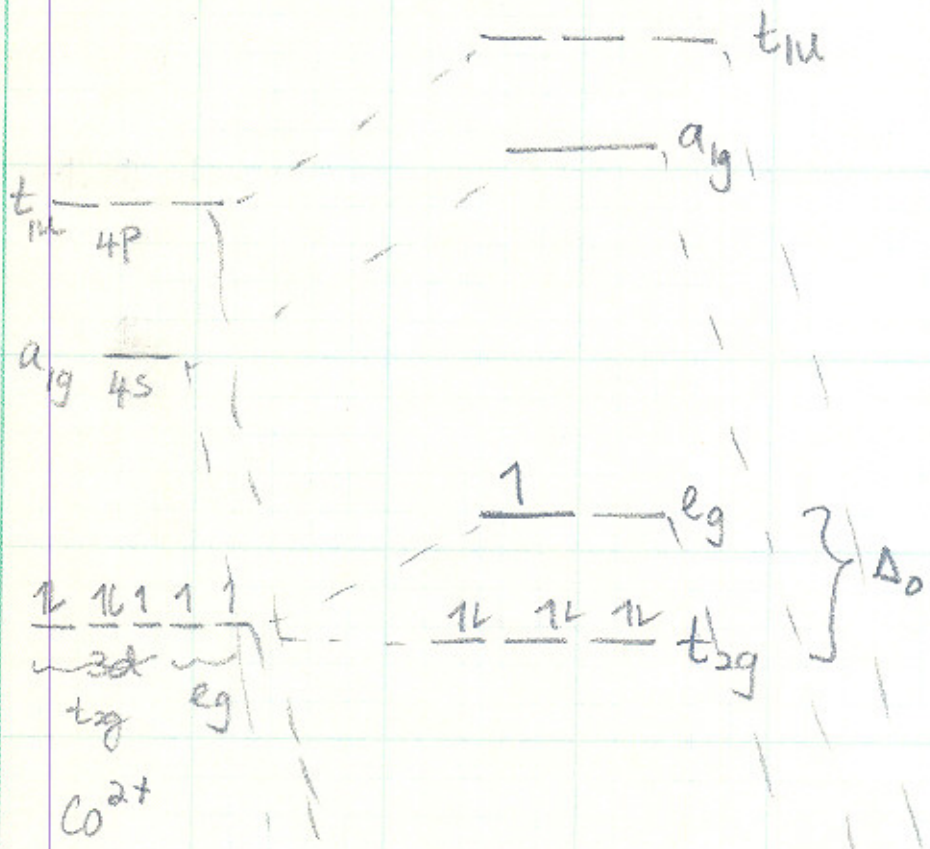
(b) allowed transitions are



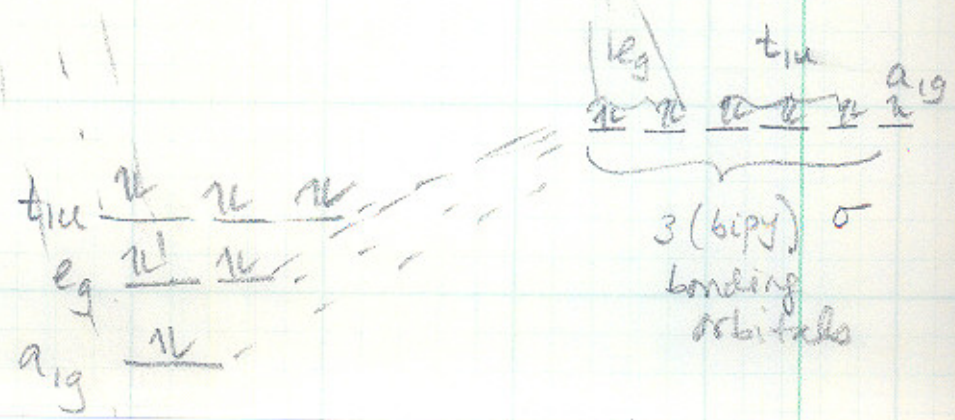
~~strong~~

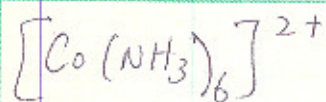
(c) $LFSE = 6 \left[-\frac{2}{5} \Delta_0 \right] + 1 \left(\frac{3}{5} \Delta_0 \right) = \underline{\underline{-\frac{9}{5} \Delta_0}}$

(d) broad bands (as in most solids).

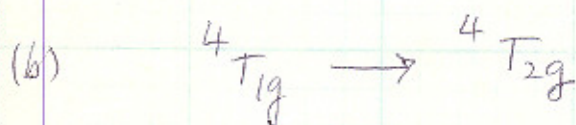
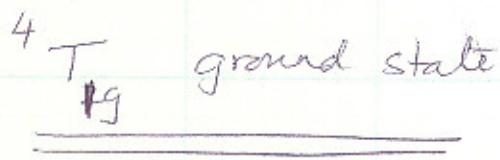
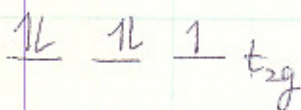
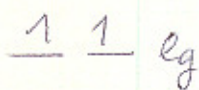
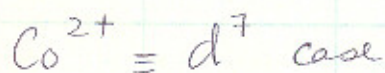


MO diagram for $[Co(bipy)_3]^{2+}$





NH_3 is a weak field ligand & it is neutral.



} may switch or be very close in energy.

$$(c) \quad LFSE = 5\left(-\frac{2}{5}\Delta_0\right) + 2\left(\frac{3}{5}\Delta_0\right) = \underline{\underline{-\frac{4}{5}\Delta_0}}$$

(d) broad, like most solids.

The MO Diagram is very similar to that of

