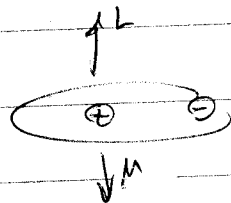
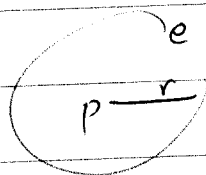


Finishing modern physics - week 8  
 Angular momentum, magnetic moment, Zeeman effect

BOHR ATOM

$$F = ma$$



24 Ex 9  
697

$$\frac{kqQ}{r^2} = \frac{e^2}{4\pi\epsilon_0 r^2} = \frac{mv^2}{r} \rightarrow \text{Solve for } v$$

Current  $I = \frac{dq}{dt} = \frac{e}{\text{period}}$

period  $= T = \frac{\text{circumference}}{v} = \frac{2\pi r}{v}$

$$I = \frac{ev}{2\pi r}$$

Area of  $e^-$  orbit  $A = \pi r^2$

MAGNETIC MOMENT  $\mu \equiv IA = \frac{ev}{2\pi r} \pi r^2 = \frac{evr}{2}$

NO. 7  
1015

Write this using ANGULAR MOMENTUM  $L = mvr \rightarrow vr = \frac{L}{m_e}$   
 solve for  $vr$ , plug into  $\mu = \frac{e}{2} \frac{L}{m_e}$

QUANTIZATION of orbital angular momentum:  $L = n \frac{h}{2\pi}$

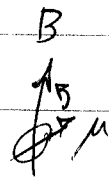
Write  $\mu$  in terms of  $m_e$ :  $\mu = \frac{e}{2} \frac{m_e \hbar}{m_e} = -m_B / m_B$   
 new quantum  $\hbar$

BOHR MAGNETON  $\mu_B = \frac{e\hbar}{2m_e}$

Bohr atom: electron's orbit  $\rightarrow$  magnetic dipole  
(DRAW B FIELD)



Magnetic dipole experiences TORQUE in a magnetic field



Potential energy  $U = -\vec{\mu} \cdot \vec{B}$

Zeeman effect: energy levels SPLIT by  $\Delta U = \mu_B B$

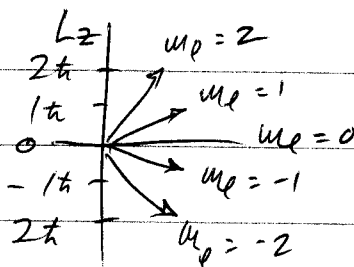
$\frac{40-2}{1005}$  Orbital angular momentum QUANTUM NUMBER  $l = 0, 1, \dots, (n-1)$

magnetic quantum number  $m_l$  indicates DIRECTION of  $l$

Orbital angular momentum  $L = \sqrt{l(l+1)} \hbar$

For  $B_z$ ,  $L_z = m_l \hbar$

If  $l = 2$ , then  $m_l$  can be 0,  $\pm 1$ , or  $\pm 2$ .



$l=0$	$l=1$	$l=2$	$l=3$
s	p	d	f
0	1	2	3

MNEMONIC for

Filling order:

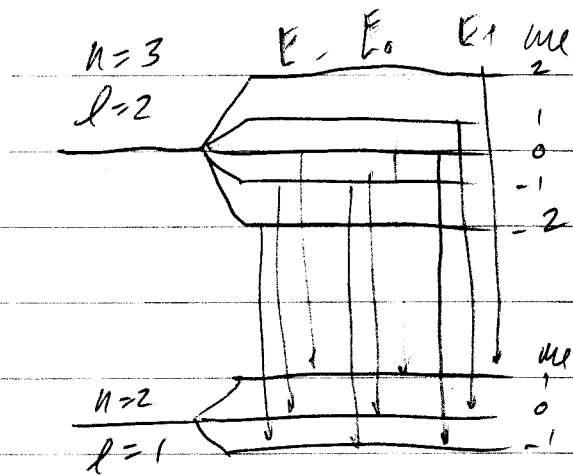
s0, p1, p0, d2, p1, s0

$l=0$	$l=1$	$l=2$	$l=3$
-------	-------	-------	-------

$$\mu_B = \frac{e\hbar}{2m_e} = \frac{1.6 \times 10^{-19} \times 1.05 \times 10^{-34}}{2 \times 9.11 \times 10^{-31}} \text{ kg} = 9.266 \times 10^{-24} \frac{\text{J}}{\text{T}}$$

45. (I) Verify that the Bohr magneton has the value  $\mu_B = 9.27 \times 10^{-24} \text{ J/T}$  (see Eq. 40-12).

46. (II) Suppose that the splitting of energy levels shown in Fig. 40-4 was produced by a 2.0-T magnetic field. (a) What is the separation in energy between adjacent  $m_l$  levels for the same  $l$ ? (b) How many different wavelengths will there be for 3d to 2p transitions, if  $m_l$  can change only by  $\pm 1$  or 0? (c) What is the wavelength for each of these transitions?



a)  $U = -\mu_z B = -\mu_B m_l B$

$$\Delta U = \Delta m_l \mu_B B = 1.2 \times 10^{-4} \text{ eV}$$

b) Transitions are allowed from  $\Delta m_l = 0, \pm 1$

c)  $\frac{1}{\lambda_0} = R \left[ \frac{1}{n^2} - \frac{1}{n'^2} \right] = 1.0974 \times 10^7 \frac{1}{\text{m}} \left( \frac{1}{4} - \frac{1}{9} \right) \quad n=2, n'=3$   
 $\lambda_0 = 656.10 \text{ nm}$

$\frac{40-2}{1005}$

$$E = \frac{hc}{\lambda}, \quad \frac{dE}{d\lambda} = hc \frac{d}{d\lambda} (\lambda^{-1}) = hc \left( -\frac{1}{\lambda^2} \right)$$

$$\frac{\Delta E}{E} = -\frac{\Delta \lambda}{\lambda} \Rightarrow \Delta E = -\frac{hc}{\lambda} \frac{\Delta \lambda}{\lambda} = \mu_B B = 9.27 \times 10^{-24} \frac{\text{J}}{\text{T}} (2\text{T})$$

$$\Delta E = 1.85 \times 10^{-23} \text{ J} = 1.16 \times 10^{-4} \text{ eV}$$

$$E_0 = \frac{hc}{\lambda_0} = \frac{6.63 \times 10^{-34} \text{ J}\cdot\text{s}}{656.10 \times 10^{-9} \text{ m}} \times \frac{3 \times 10^8 \text{ m}}{\text{s}} = 3.03 \times 10^{-19} \text{ J} \cdot \mu = 1.895 \text{ eV}$$

$$E_0 + \Delta E = E_+ = \frac{hc}{\lambda_0 - \Delta \lambda} = \frac{hc}{\lambda_+} \quad \text{or} \quad \frac{\Delta \lambda}{\lambda_0} = \frac{\Delta E}{E} = 6.12 \times 10^{-5} \rightarrow \lambda_+ = \left( 1 + \frac{\Delta \lambda}{\lambda_0} \right) \lambda_0 = 656.14 \text{ nm}$$

$$E_0 - \Delta E = E_- = \frac{hc}{\lambda_0 + \Delta \lambda} = \frac{hc}{\lambda_-} \quad \lambda_- = \left( 1 - \frac{\Delta \lambda}{\lambda_0} \right) \lambda_0 = 656.06 \text{ nm}$$

39-3  
981

# Heisenberg Uncertainty Principle

Uncertainty in position  $\Delta x \approx \lambda$  of light used to look at particle.

Photon can change particle's momentum by  $\Delta p \approx \frac{h}{\lambda}$

$$\Delta x \Delta p \approx \underline{\hspace{2cm}}$$

$$\Delta x \Delta p \approx \hbar = \frac{h}{2\pi} \quad \text{Do \# 7}$$

The more precisely we measure position (with light of small  $\lambda$ ) the less precisely we can know momentum.

Time uncertainty  $\Delta t \approx \frac{\Delta x}{c} \approx \frac{\lambda}{c}$  (travel time of photon across object)

Photon can change particle's energy by  $\Delta E \approx \frac{hc}{\lambda}$

$$\Delta E \Delta t \approx \underline{\hspace{2cm}}$$

$$\Delta E \Delta t \approx \hbar$$

$$\text{Do \# 9}$$