

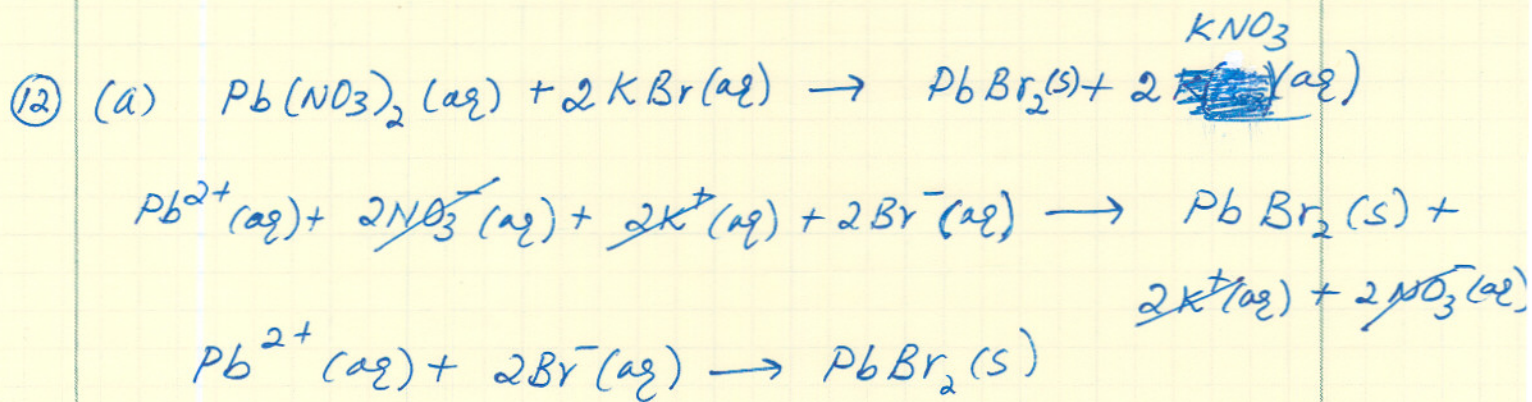
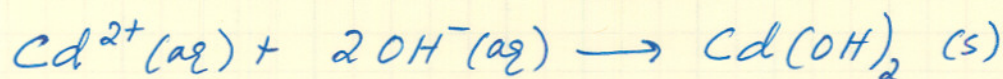
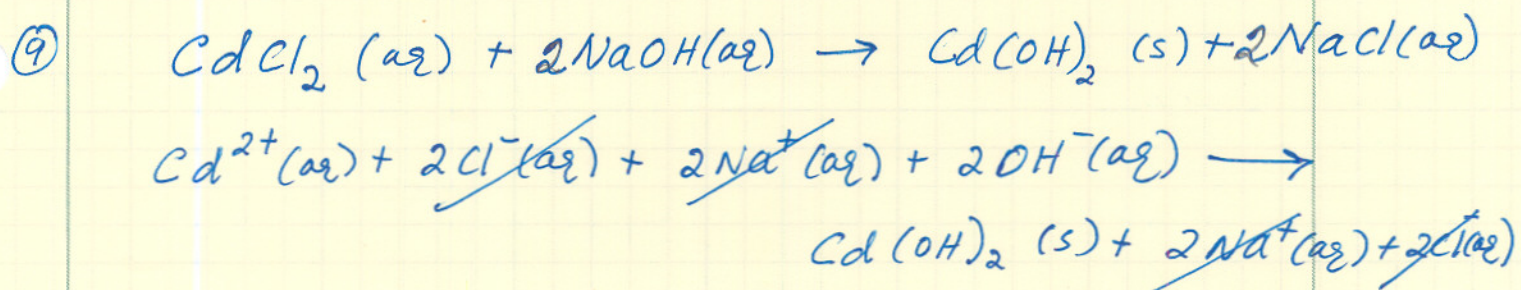
INTRODUCTION TO NATURAL SCIENCE

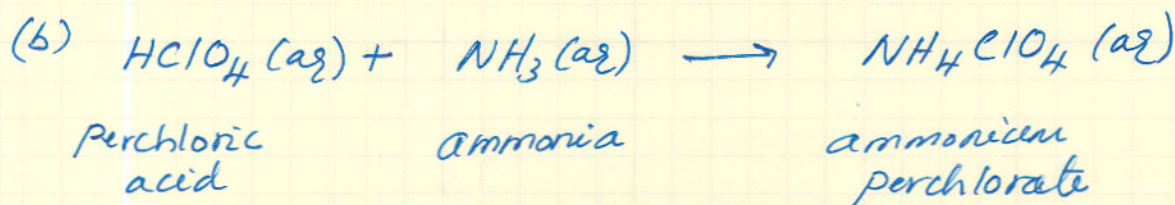
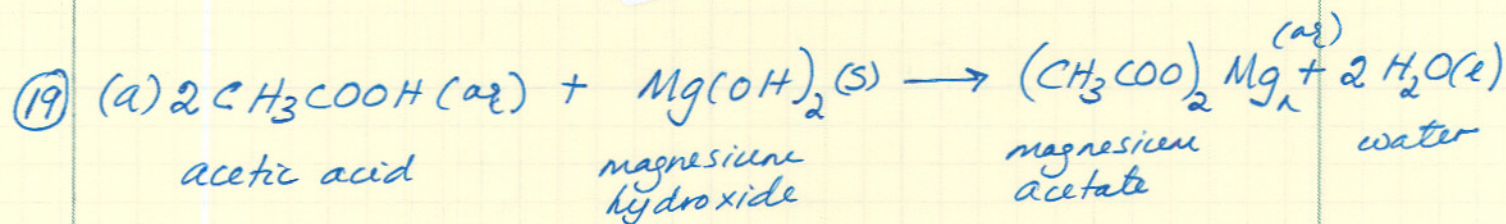
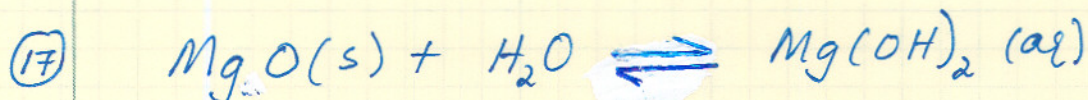
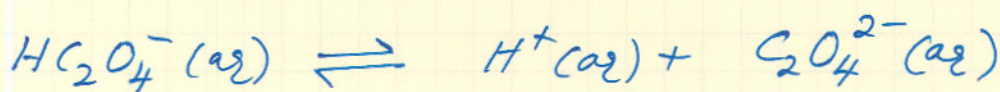
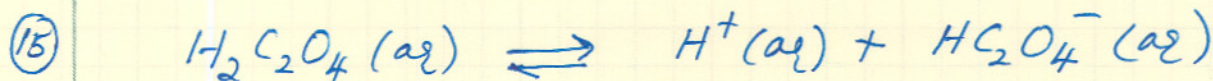
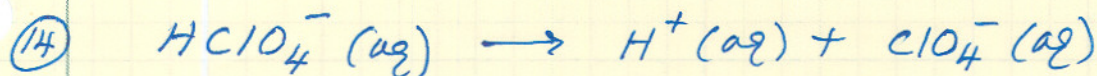
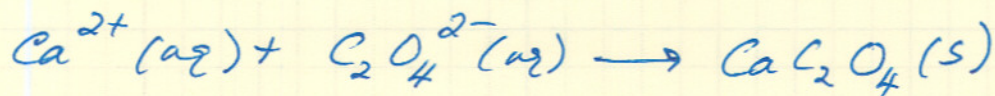
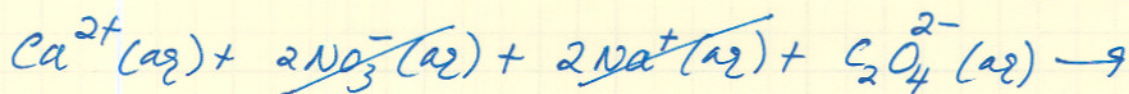
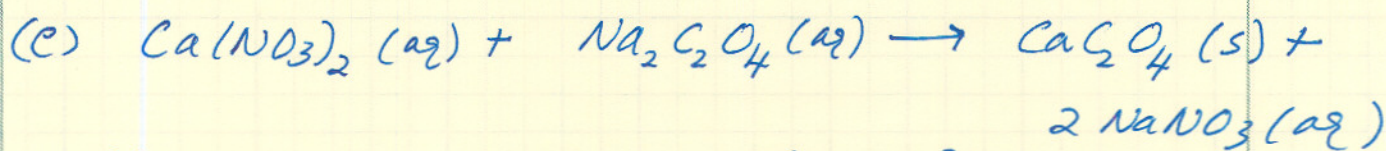
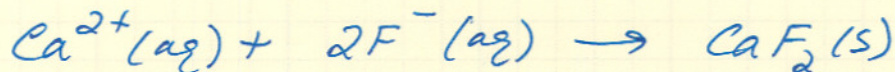
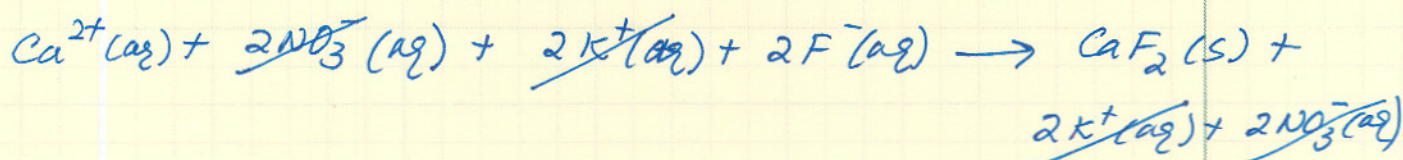
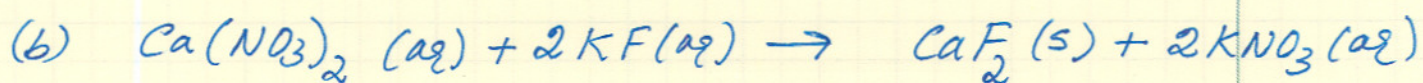
CHEMISTRY HW - FALL 2006 - WEEK 6

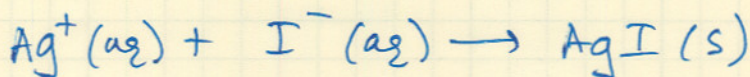
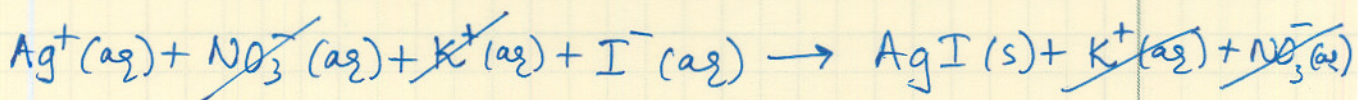
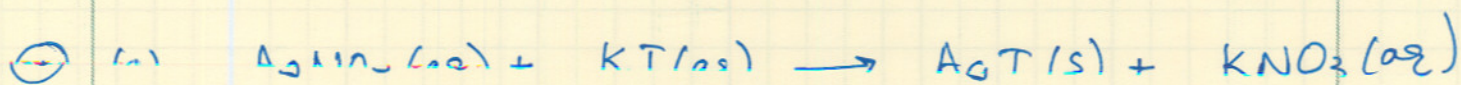
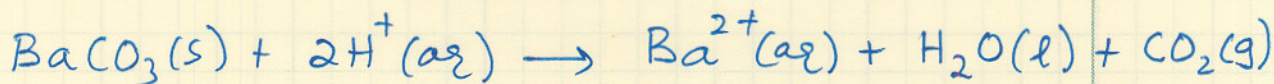
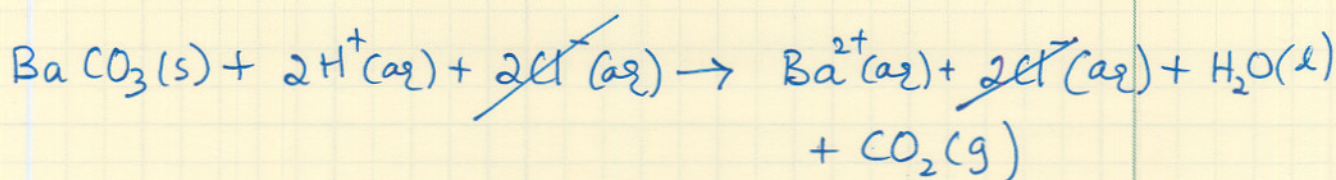
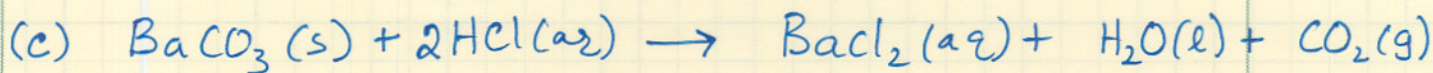
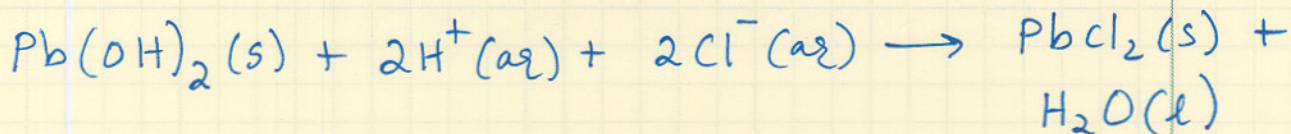
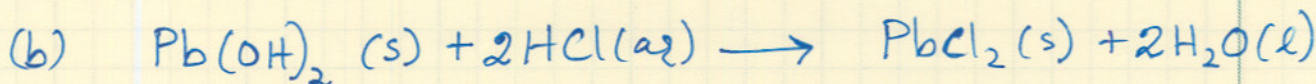
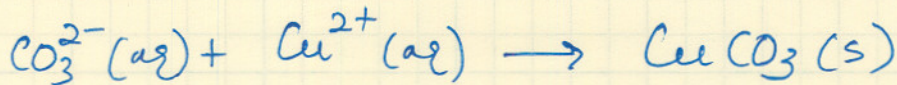
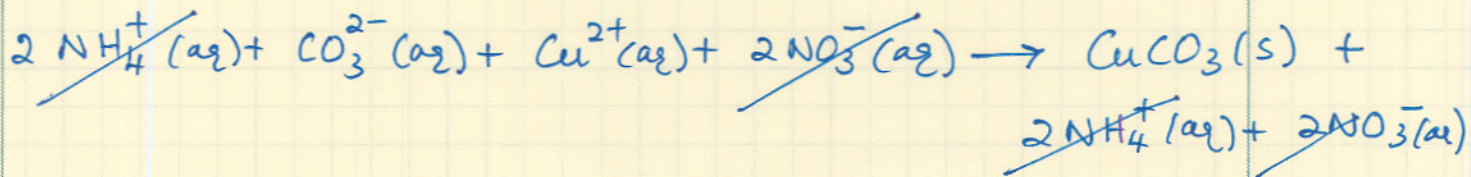
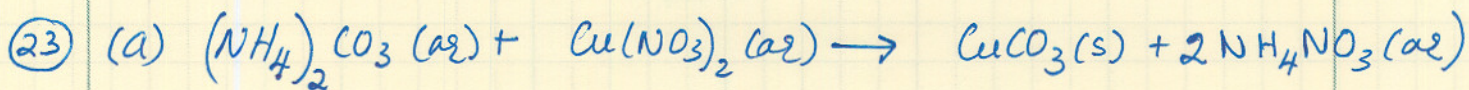
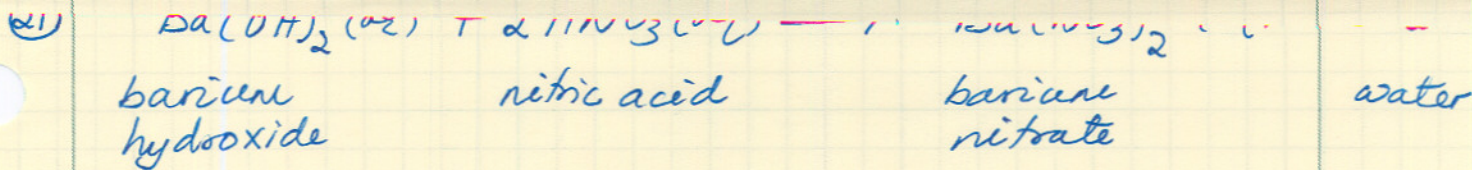
Chapter 5

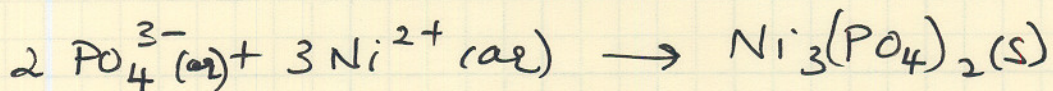
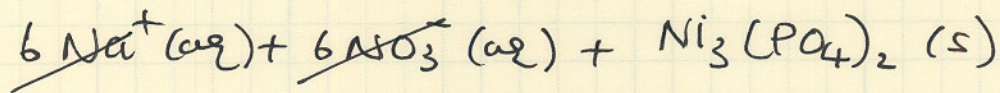
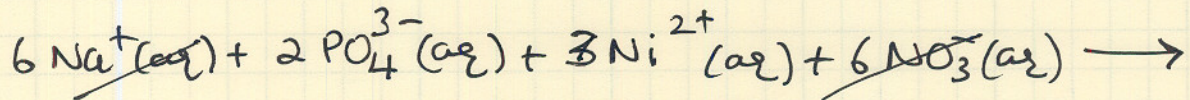
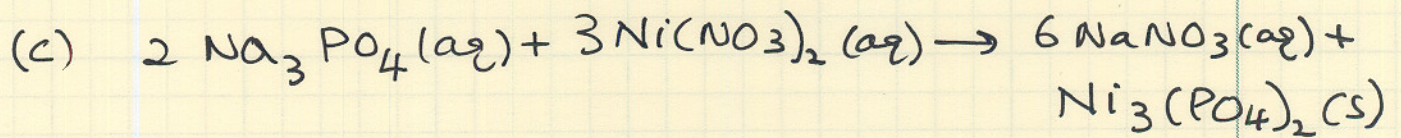
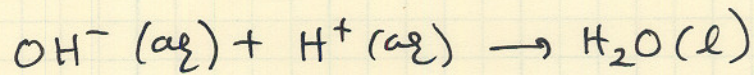
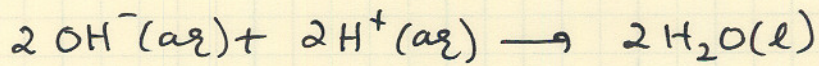
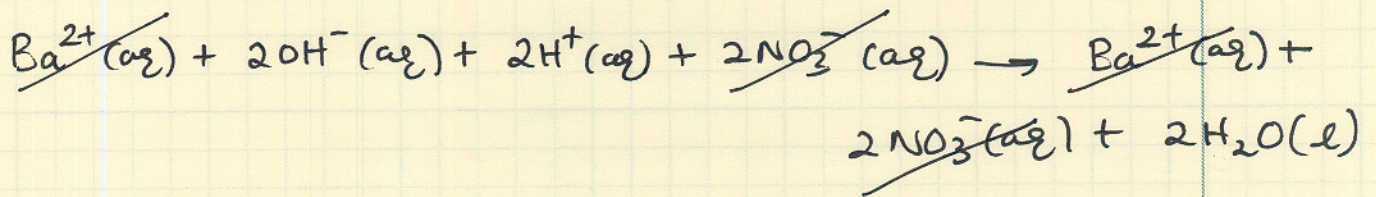
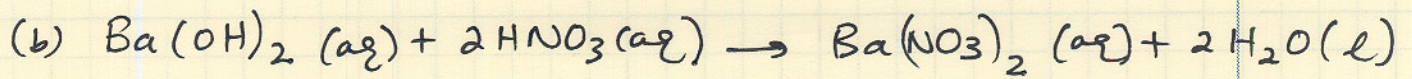
- ⑤ (a) K^+ and OH^- (b) K^+ and SO_4^{2-} (c) Li^+ and NO_3^-
(d) NH_4^+ and SO_4^{2-}

- ⑦ (a) Na_2CO_3 is water soluble. Na^+ , CO_3^{2-}
(b) $CuSO_4$ is water soluble Cu^{2+} , SO_4^{2-}
(c) NiS not soluble
(d) $BaBr_2$ is soluble. Ba^{2+} , Br^-









2